## **IN THE CLAIMS:**

Claims 1 - 22 (Canceled)

- 23. (Previously presented) A method of producing a positive electrode active material for a non-aqueos electrolyte cell, comprising the sequential steps of:
  - (1) mixing ingredients of a lithium composite manganese oxide;
  - (2) molding the mixture under pressure; and
- (3) sintering the molded mixture at a temperature not lower than 600°C and not higher than 850°C, wherein
- (a) the positive electrode active material comprises lithium composite manganese oxide having a spinel structure whose primary particle diameter is not less than 0.05  $\mu$ m and not greater than 10  $\mu$ m forming an aggregate, and whose specific surface area measured by the BET method is not less than 0.2 m²/g and not greater than 2 m²/g;
- (b) the non-aqueos electrolyte cell comprises a negative electrode having a material capable of reversively doping and dedoping lithium, wherein the material capable of reversively doping and dedoping lithium is at least one selected from the group consisting of a carbon material, metal lithium, lithium alloy, polyacene, and polypyrol;
- (c) wherein the carbon material is at least one selected from the group consisting of pyrocarbon, coke, glassy carbon, organic polymer compound sintered body, and carbon fiber; and
- (d) wherein the lithium composite manganese oxide is expressed by a general formula  $Li_xMn_{2-y}M_yO_4$ , wherein .09  $\leq$  x  $\leq$  1.4; 0<y<.3; and M is one or more materials selected from the group consisting of Ti, V, Cr, Fe, Co, Ni, and Al.
- 24. (Previously Presented) The method of claim 23, wherein the spinel structure has a primary particle diameter that is not less than 0.1  $\mu$ m and not greater than 5  $\mu$ m.
- 25. (Previously Presented) The method of claim 23, wherein the spinel structure has a primary particle diameter that is not less than 0.5  $\mu$ m and not greater than 3  $\mu$ m.

- 26. (Previously Presented) The method of claim 23, further comprising pulverizing the sintered mixture.
- 27. (Previously Presented) The method of claim 23, wherein the non-aqueous electrolyte cell comprises a positive electrode including graphite and binder in addition to the positive electrode active material.
- 28. (Previously Presented) The method of claim 27, wherein the positive electrode comprising a metal foil and a positive electrode composite agent on said metal foil, said positive electrode composite agent containing the positive electrode active material and the binder dissolved in a solvent to create a slurry uniformly applied on said metal foil and dried.

Claims 29-34 (Canceled)

- 35. (Previously presented) A method of producing a positive electrode active material for a non-aqueos electrolyte cell, comprising the sequence steps of:
  - (1) mixing ingredients of a lithium composite manganese oxide;
  - (2) molding the mixture under pressure; and
- (3) sintering the molded mixture at a temperature not lower than 600°C and not higher than 850°C, wherein the positive electrode active material comprises lithium composite manganese oxide having a spinel structure whose primary particle diameter is not less than 0.05  $\mu$ m and not greater than 10  $\mu$ m forming an aggregate, and whose specific surface area measured by the BET method is not less than 0.2 m²/g and not greater than 2 m²/g.